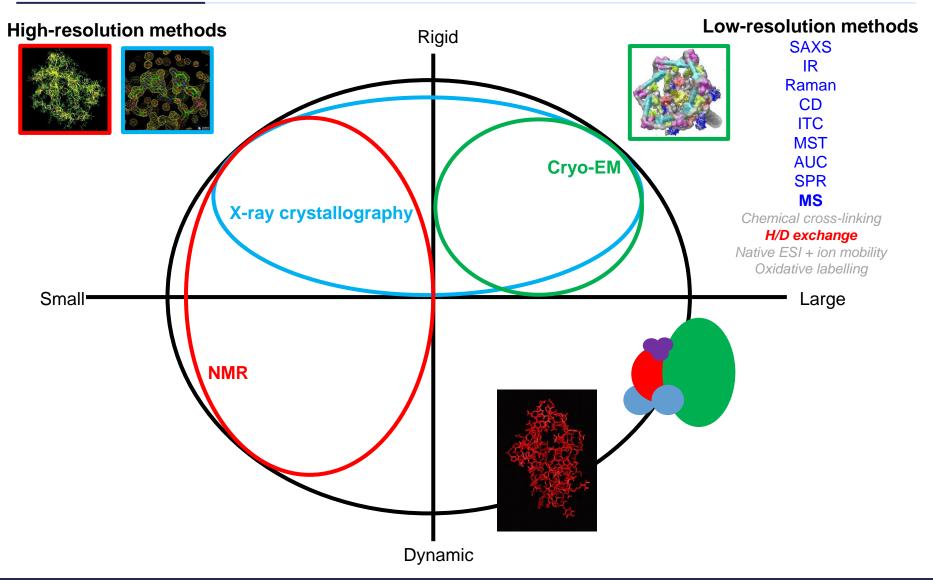
Structural mass spectrometry hydrogen/deuterium exchange

Petr Man

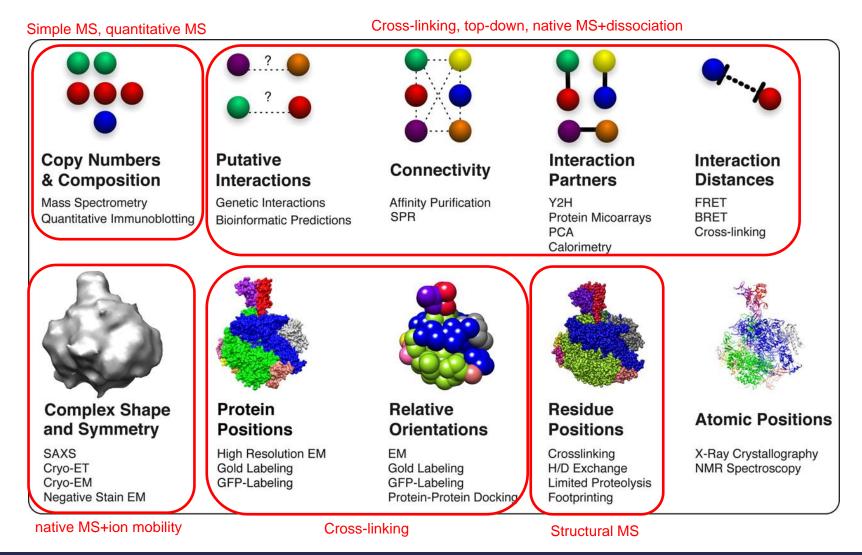
Structural Biology and Cell Signalling Institute of Microbiology, Czech Academy of Sciences

Structural biology methods



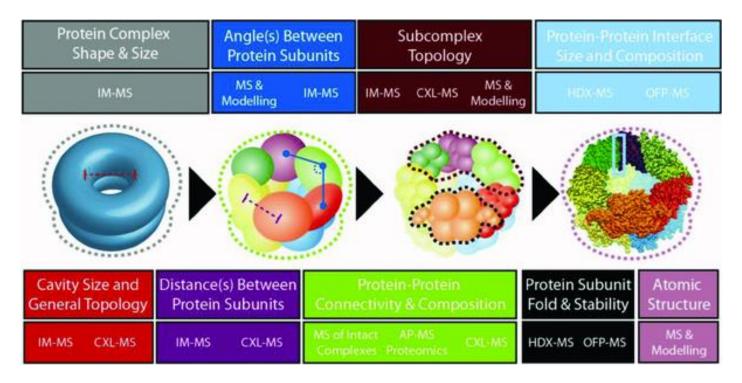


Structural biology approaches



MS

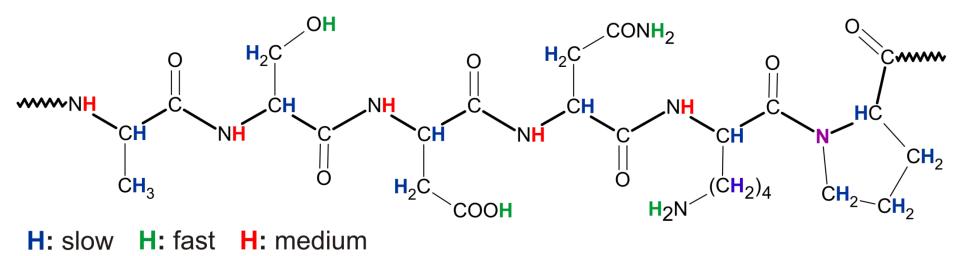
What can we get using mass spectrometry



- IM ion mobility
- CXL chemical cross-linking
- AP afinity purification
- *OFP oxidative footprinting*
- HDX hydrogen/deuterium exchange



ISOTOPE EXCHANGE IN PROTEINS ~ Ala – Ser – Asp – Asn – Lys – Pro ~



³H Tritium

				_	
	¹ H	² H	³ H		•
occurence [%]	99.988 5	0.0115	trace		
				¹ H Hydrogen-1	² H Deuteriu



...Kaj Ulrik Linderstrøm-Lang

SHORT COMMUNICATIONS, PRELIMINARY NOTES

VOL. 14 (1954)

EXCHANGE OF HYDROGEN ATOMS IN INSULIN WITH DEUTERIUM ATOMS IN AQUEOUS SOLUTIONS

> by AASE HVIDT AND K. LINDERSTRØM-LANG * Carlsberg Laboratorium, Copenhagen (Denmark)

"Cartesian diver" Proteins are migrating in tubes with density gradient until they stop at the point where the densities are equal

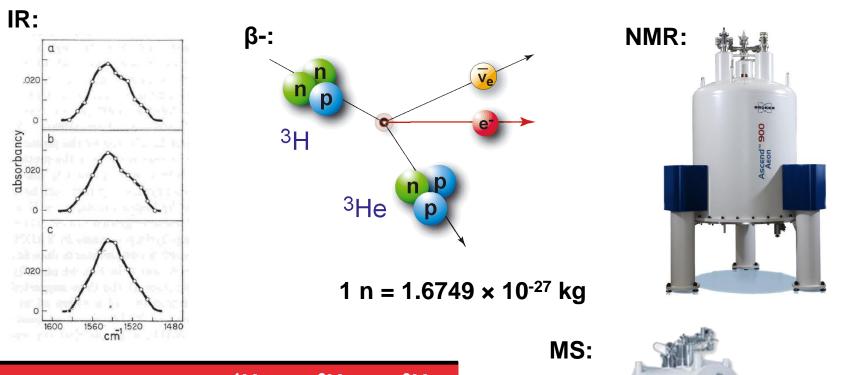
	¹ H	² H	³ Н
%	99.9885	0.0115	trace
density [g/cm ³]	1.000	1.106	1.215





574

Methods of detection

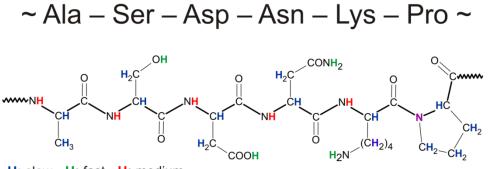


	¹ H	² H	³ H
%	99.9885	0.0115	trace
density [g/cm ³]	1.000	1.106	1.215
spin	1⁄2+	1+	1⁄2+
mass [u]	1.00783	2.01410	3.01605





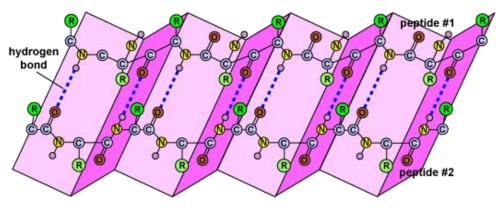
Factors affecting H/D exchange



H: slow H: fast H: medium

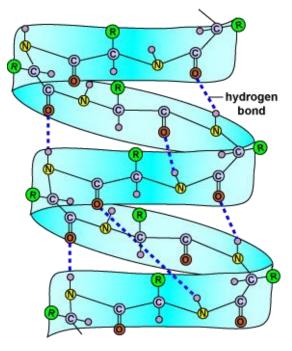
hydrogen bonding

solvent accessibility



beta pleated sheet

alpha helix

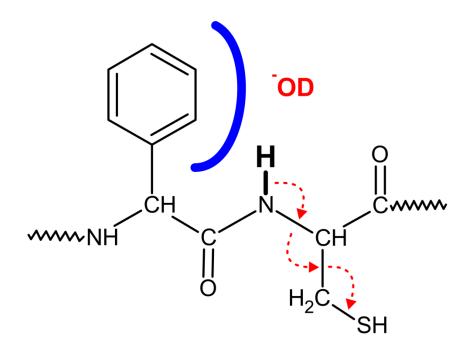




Factors affecting H/D exchange

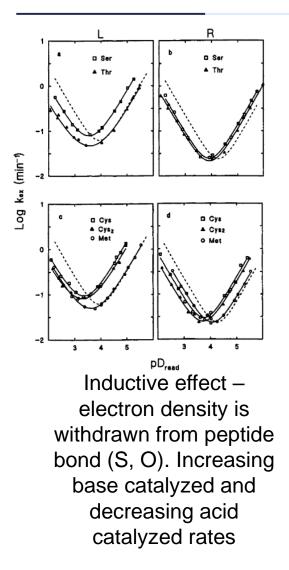
Side chains

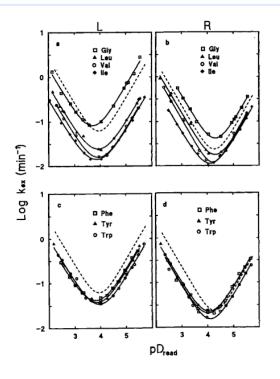
(acidity, steric shielding)





Factors affecting H/D exchange – side chain effects





Downward shift due to steric hindrance effect of aliphatic and aromartic side chains. Aromatics also shows inductive effect

TABLE II. Effects of Amino Acid Side Chains on the HX Rates of Neighboring Peptides*

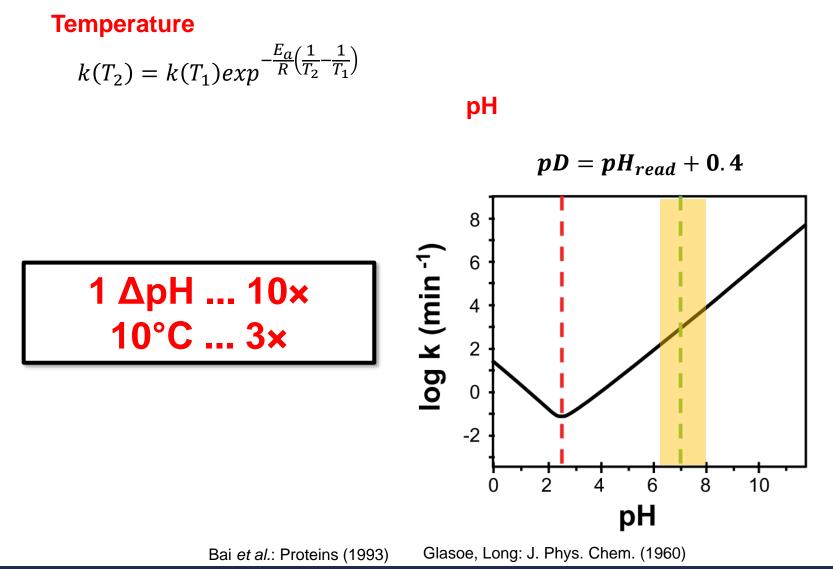
	$Logk_{ex}(X) - Logk_{ex}(Ala)$			
	Acid		Base	
	catal	ysis	catalysis	
Side chain (X)	L	R	L	R
Ala	0.00	0.00	0.00	0.00
Arg	-0.59	-0.32	0.08	0.22
Asn	-0.58	-0.13	0.49	0.32
$Asp(COO^{-})$	(0.9)	0.58	-0.30	-0.18
Asp(COOH)	(-0.9)	-0.12	0.69	(0.6)
Cys	-0.54	-0.46	0.62	0.55
Cys ₂	-0.74	-0.58	0.55	0.46
Gly	-0.22	0.22	0.27	0.17
Gln	-0.47	-0.27	0.06	0.20
Glu(COO ⁻)	(-0.9)	0.31	-0.51	-0.15
Glu(COOH)	(-0.6)	-0.27	0.24	0.39
His			-0.10	0.14
His ⁺	(-0.8)	-0.51	(0.8)	0.83
Ile	-0.91	-0.59	-0.73	-0.23
Leu	-0.57	-0.13	-0.58	-0.21
Lys	-0.56	-0.29	-0.04	0.12
Met	-0.64	-0.28	-0.01	0.11
Phe	-0.52	-0.43	-0.24	0.06
Pro(trans)		-0.19		-0.24
Pro(cis)		-0.85		0.60
Ser	-0.44	-0.39	0.37	0.30
Thr	-0.79	-0.47	-0.07	0.20
Trp	-0.40	-0.44	-0.41	-0.11
Tyr	-0.41	-0.37	-0.27	0.05
Val	-0.74	-0.30	-0.70	-0.14
N-term (NH ₃ ⁺)		-1.32		1.62
C-term (COO ⁻)	0.96		(-1.8)	
C-term (COOH)	(0.05)			

*L and R refer to peptide groups to the left and right, respectively, of the side chain indicated (replacing the λ and ρ terminology of Molday et al.⁶). Values are listed in logarithmic form for use with Eq. (2). Values in parentheses are less well determined.

Bai et al.: Proteins (1993)



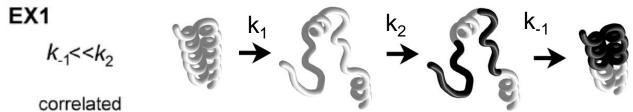
Factors affecting H/D exchange





H/D EXCHANGE KINETICS

 $\mathbf{F}_{\mathbf{H}} \underset{k_{1}}{\overset{\kappa_{1}}{\longleftrightarrow}} \mathbf{U}_{\mathbf{H}} \underset{D_{2}O}{\overset{k_{2}}{\longrightarrow}} \mathbf{U}_{\mathbf{D}} \underset{k_{1}}{\overset{\kappa_{1}}{\longleftrightarrow}} \mathbf{F}_{\mathbf{D}}$ EX2 $\stackrel{k_1}{\rightarrow} \underbrace{\begin{array}{c} \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \end{array} \stackrel{k_2}{\rightarrow} \underbrace{\begin{array}{c} \\ \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \begin{array}{c} \\ \end{array} \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \end{array} \end{array} \begin{array}{c} \\ \end{array} \end{array} \end{array} \end{array} \end{array}$ \end{array} $k_{1} > k_{2}$ uncorrelated

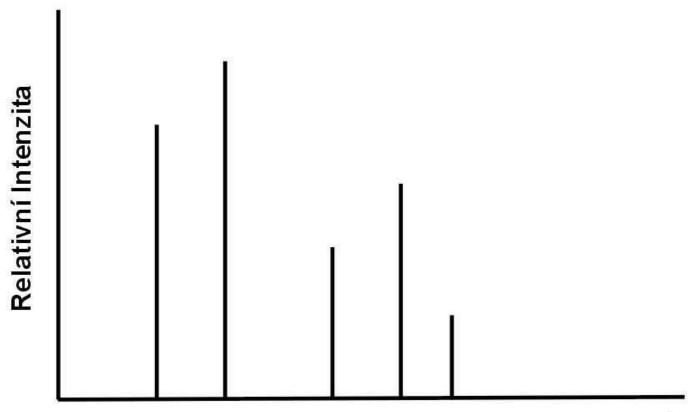












m/z

Isotope pattern

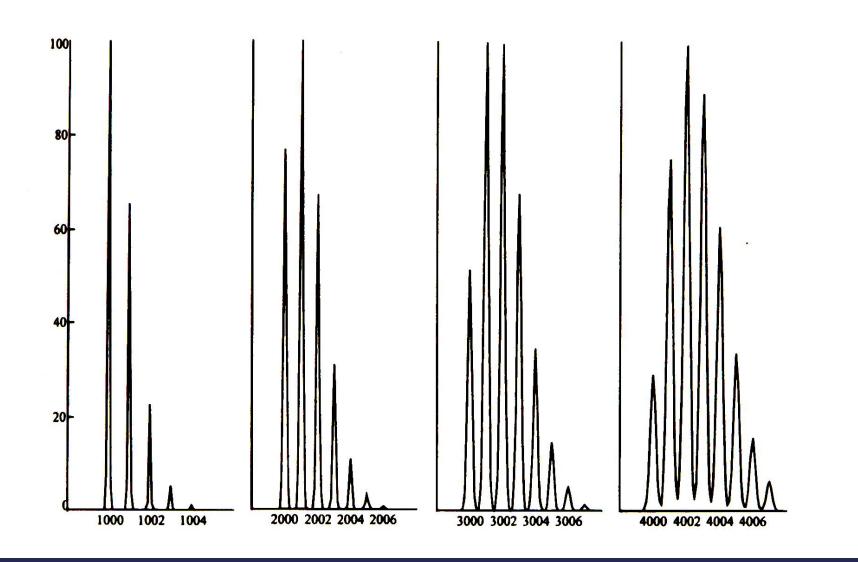
Mass spectrum in detail

12 -		
¹² C		
¹ H		2× ¹³ C
¹⁴ N		2× ² H
¹⁶ O		2× ¹⁵ N
³² S		2× ¹⁷ O
		2× ³³ S
		1× ³⁴ S
		1× ¹⁸ 0
		$1 \times {}^{13}C + 1 \times {}^{2}H$
		$1 \times {}^{13}C + 1 \times {}^{15}N$
	¹³ C	$1 \times {}^{13}C + 1 \times {}^{17}C$
	² H	$1 \times {}^{13}C + 1 \times {}^{33}S$
	¹⁵ N	$1 \times {}^{2}H + 1 \times {}^{15}N$
	¹⁷ O	$1 \times {}^{2}H + 1 \times {}^{17}C$
	³³ S	$1 \times {}^{2}H + 1 \times {}^{33}S$
		1×15 N + 1×17 C
		1×15 N + 1×33 S
		1× ¹⁷ 0 + 1× ³³ 9

Symbol	M _{nom}	M _{mono}	%
С	12	12.00000	98.9300
	13	13.00336	1.0700
н	1	1.00783	99.9885
	2	2.01411	0.1150
Ν	14	14.00307	99.6320
	15	15.00011	0.3680
0	16	15.99492	99.7570
	17	16.99913	0.0380
	18	17.99916	0.2050
S	32	31.97207	94.9300
	33	32.97146	0.7600
	34	33.96787	4.2900
	36	35.96708	0.0200
Р	31	30.97376	100.0000

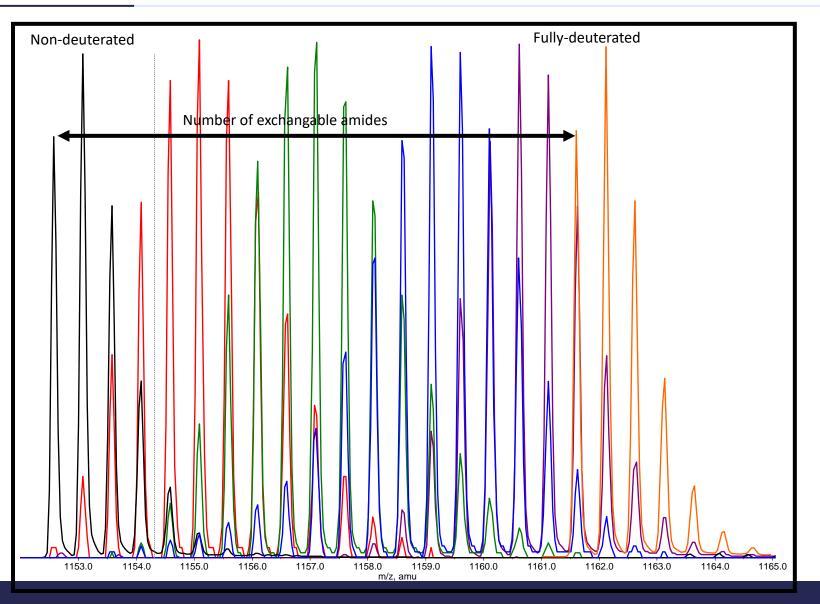


Isotope pattern at increasing mass



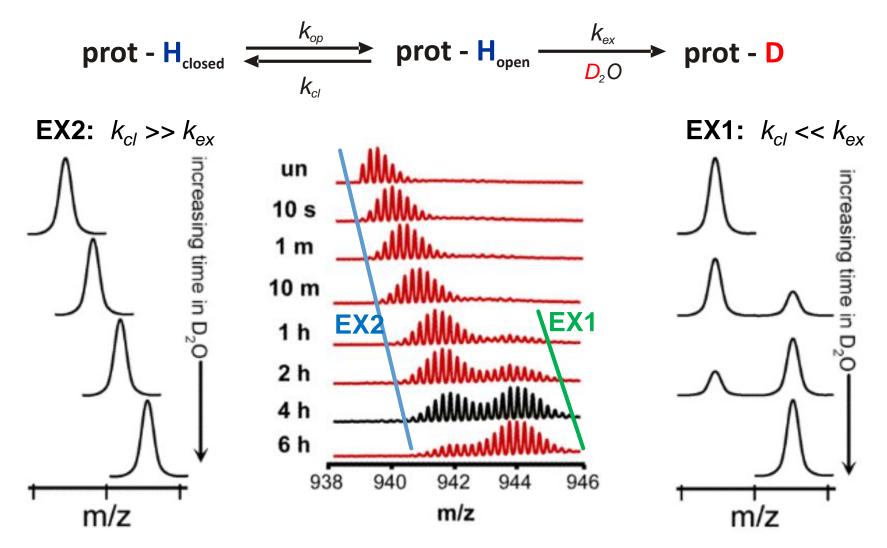


Isotope pattern changes





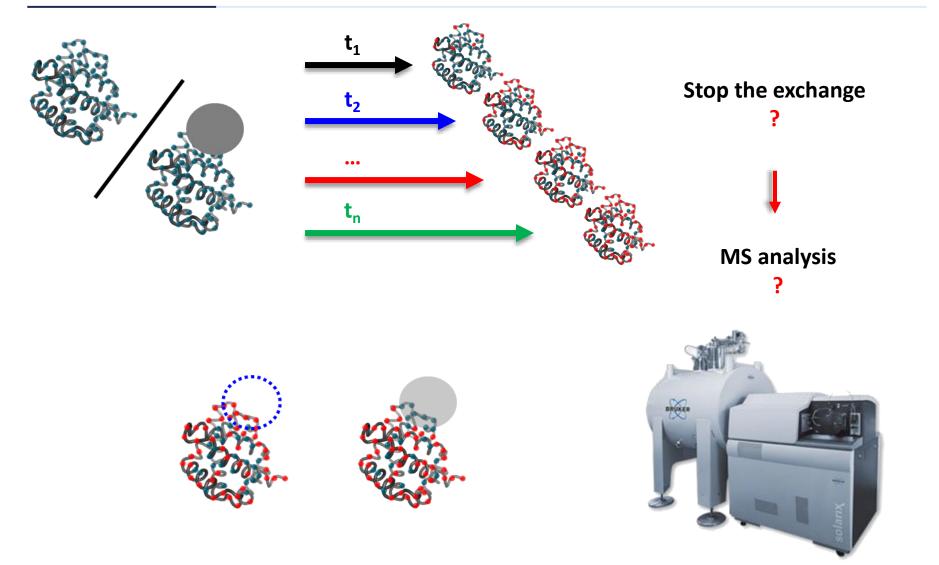
H/D EXCHANGE KINETICS



Weis DD et al.: J. Am. Soc. Mass Spectrom. (2006) Fang J et al.: Biochemistry (2011)

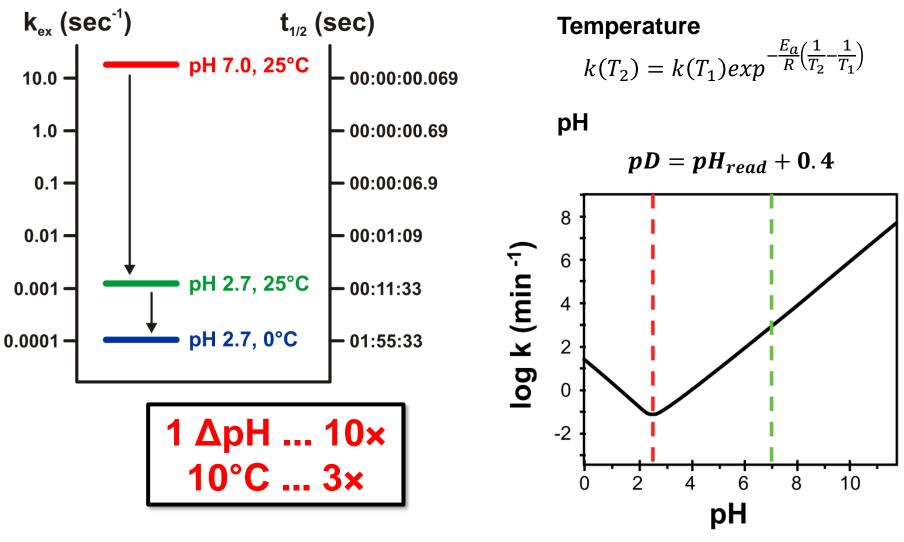


H/D exchange workflow





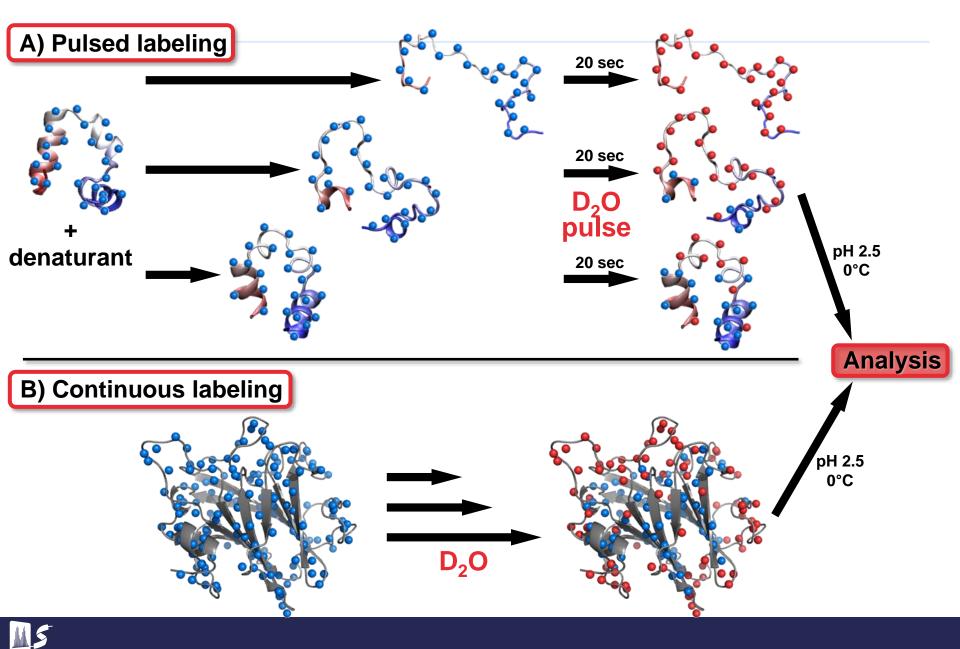
H/D exchange can be "stopped"



Bai et al.: Proteins (1993) Glasoe, Long: J. Phys. Chem. (1960) www.hxms.com

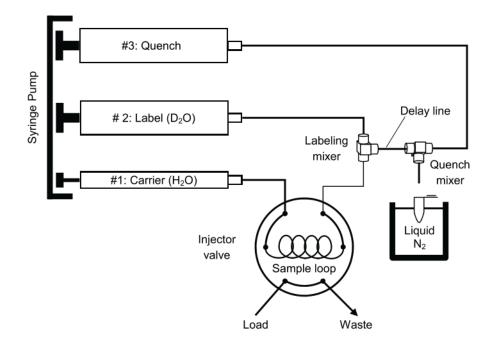


EXPERIMENTAL SETUP

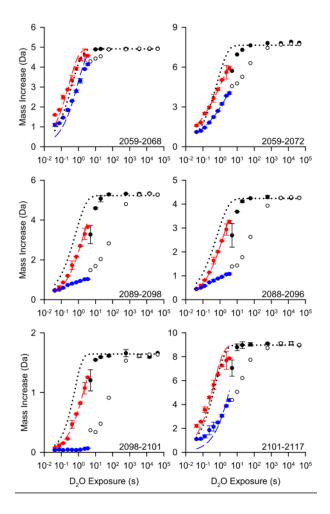


EXPERIMENTAL SETUP

C) Quenchflow – millisecond HDX

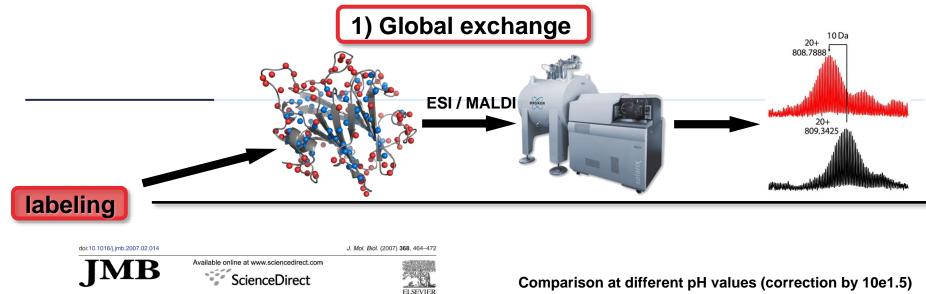


Microfluidics – mixing issues / flow



Keppel et al.: Anal Chem (2013)

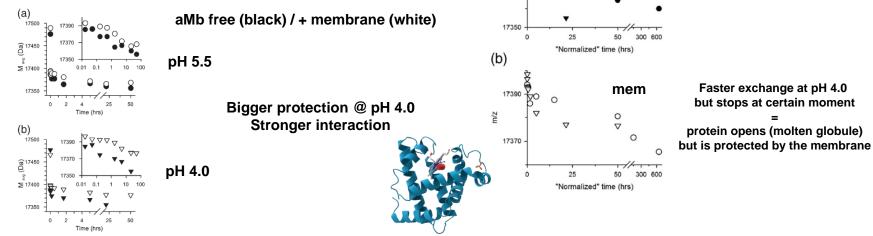




Defining the Interacting Regions between Apomyoglobin and Lipid Membrane by Hydrogen/ Deuterium Exchange Coupled to Mass Spectrometry

Petr Man¹, Caroline Montagner², Grégory Vernier², Bernard Dublet¹ Alexandre Chenal², Eric Forest^{1*} and Vincent Forge^{2*}

D/H exchange (to diminish effect of side-chains) using MALDI-TOF

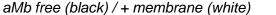


(a)

Z/m

17390

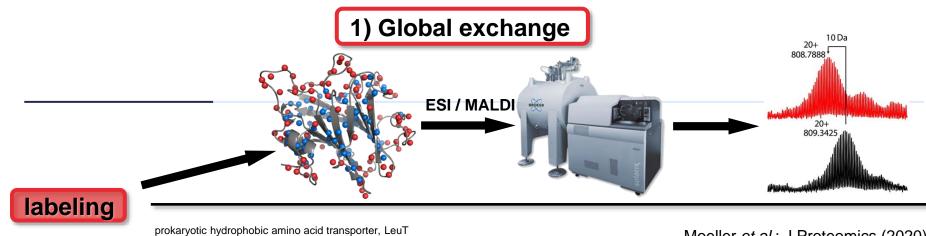
17370



sol

Faster exchange at pH 4.0

protein opens (molten globule)

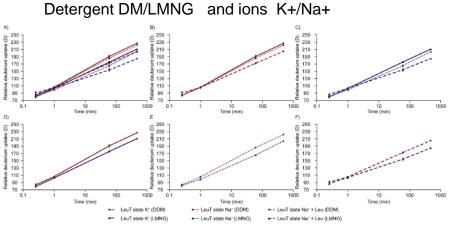


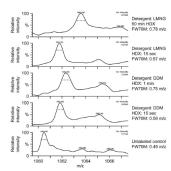


Probing the conformational impact of detergents on the integral membrane protein LeuT by global HDX-MS

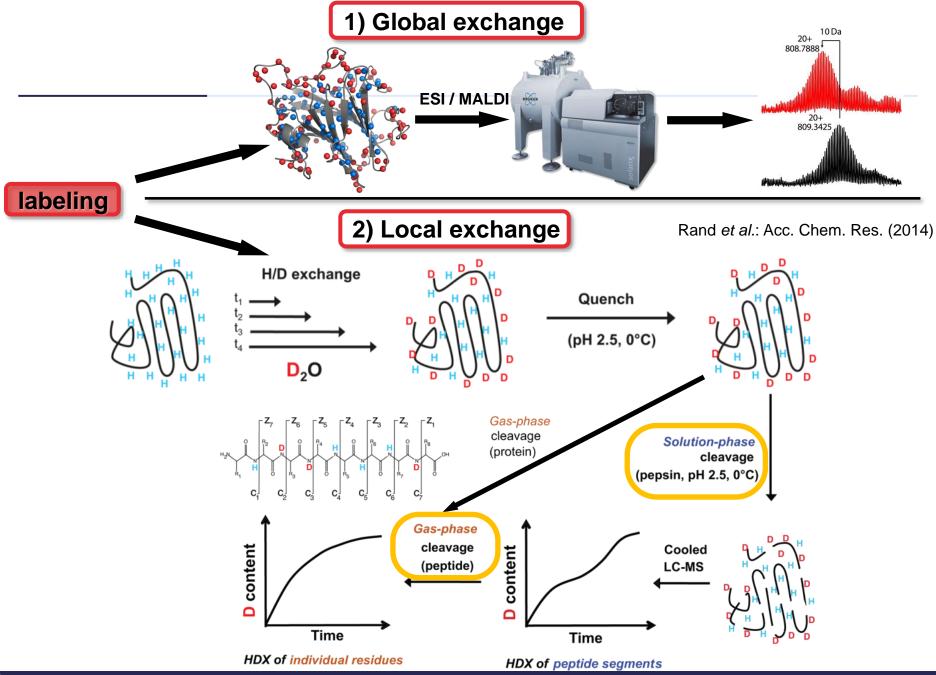
Ingvar R. Möller^{a,1}, Patrick S. Merkle^{a,1}, Dionisie Calugareanu^a, Gerard Comamala^a, Solveig Gaarde Schmidt^b, Claus J. Loland^b, Kasper D. Rand^{a,*}

Moeller et al.: J Proteomics (2020)



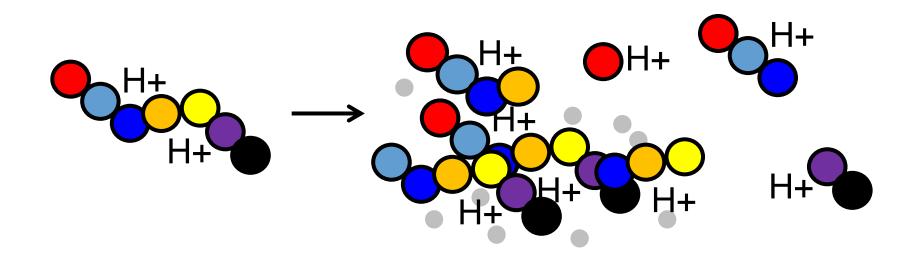


Protein dynamics (EX1/EX2) assessed through peak width



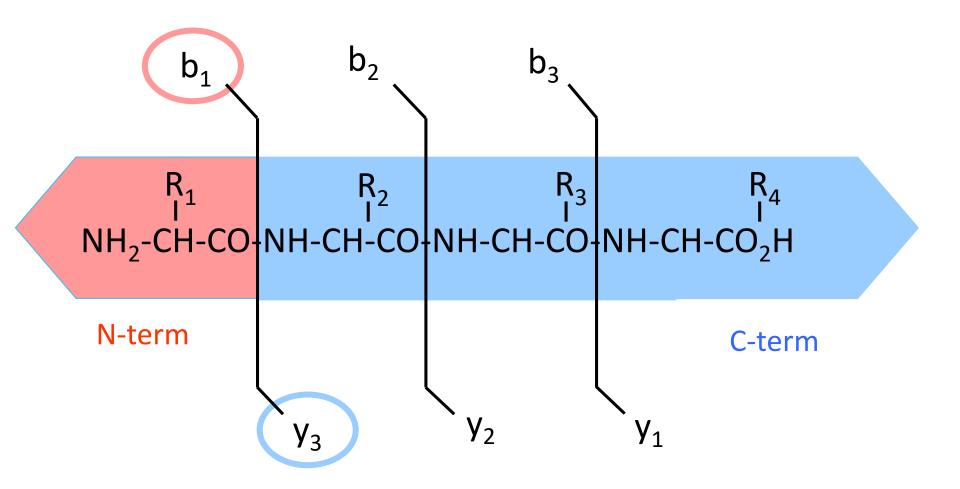


CID – collision induced dissociation





Peptide fragmentation





Peptide fragmentation

S-P	-A-F-D-S-I-M-A-E-T-I	K MH ⁺ = 1	410.6
<u>b-ions</u> +			<u>y-ions</u> +
88.1	S	PAFDSIMAETLK	1323.6
185.2	SP	- AFDSIMAETLK	1226.4
256.3	SPA	FDSIMAETLK	1155.4
403.5	SPAF	DSIMAETLK	1008.2
518.5	SPAFD	SIMAETLK	893.1
605.6	SPAFDS	IMAETLK	806.0
718.8	SPAFDSI	MAETLK	692.3
850.0	SPAFDSIM	AETLK	561.7
921.1	SPAFDSIMA	ETLK	490.6
1050.2	SPAFDSIMAE	TLK	361.5
1151.3	SPAFDSIMAET	LK	260.4
1264.4	SPAFDSIMAETL	K	147.2



Electron Capture/Transfer Dissociation (ECD/ETD)

Electron capture (E \leq 0.1 eV) - on S-S bonds and N-C_a bonds

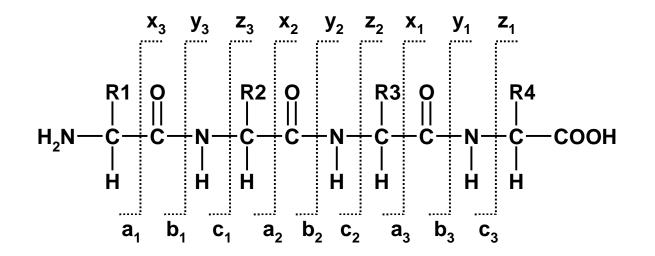
Neutralization H⁺; very fast process but requires multiply charged ions

c- and z-ion series

Electron Transfer Dissociation (ETD)

ECD not applicable in Rf field analyzers (traps, quads)

Chemical ionization = electron is transferred by an anion (anthracene, methane, ...)





UltraViolet PhotoDissociation (UVPD)

UV lasers – Nd:YAG a excimer ArF, KrF,..

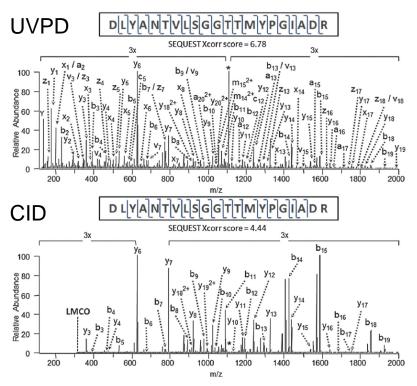
wavelength: 157nm, 193nm (peptide bond), 213nm, 266nm,...

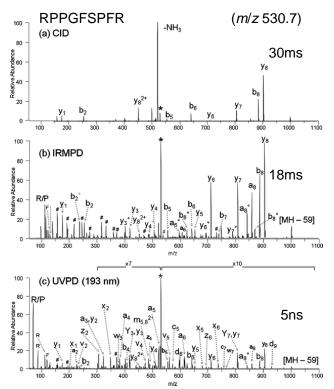
wavelength close to 280nm – fragmentation close to aromatics

Very fast fragmentation - in psec – no scrambling

Similar to ExD but no charge reduction – also applicable to 1+ ions

a/x, c/z, but also b/y ions and side chain fragments

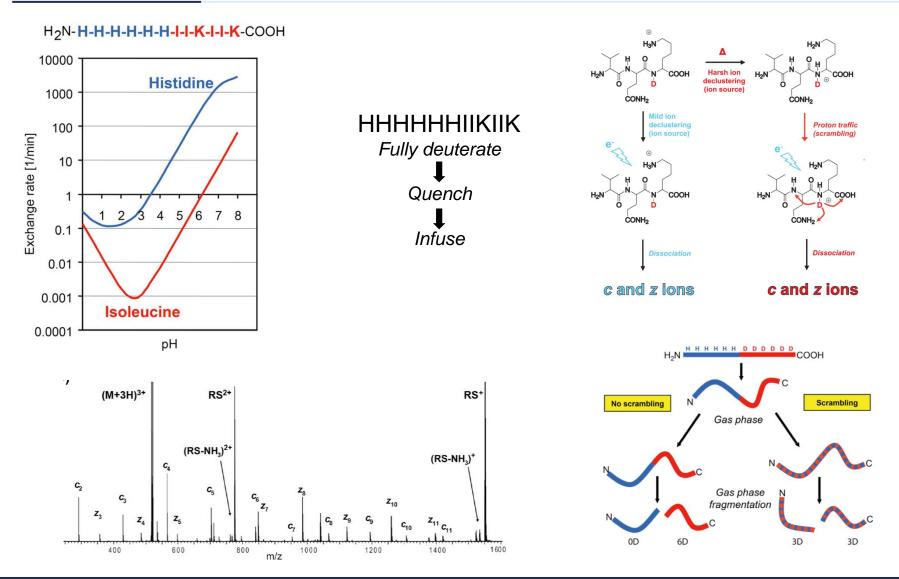






Scrambling test

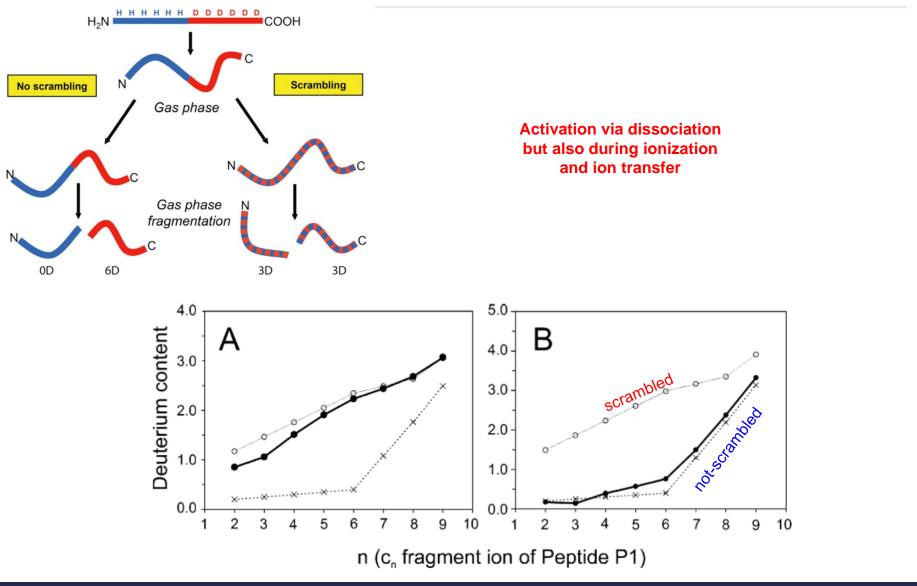
Kaltashov, Eyles: J. Mass Spectrom. (2002) Rand *et al.*: Acc. Chem. Res. (2014)



M5

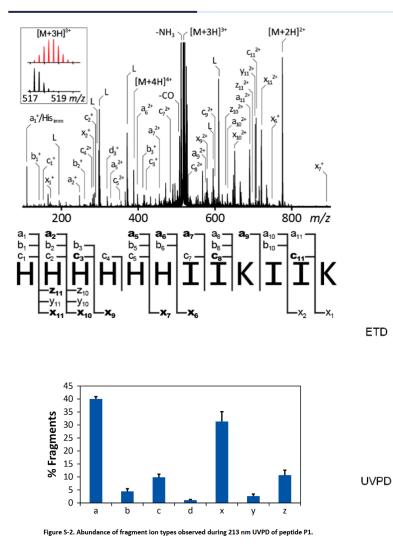
Scrambling – ETD/ECD

Kaltashov, Eyles: J. Mass Spectrom. (2002) Rand *et al.*: Acc. Chem. Res. (2014)





Scrambling - UVPD



ETD

Mistraz U.H. et al Anal Chem 2018 Brodie N.I. et al Anal Chem 2018

UVPD is applicable (similarly to ExD) More fragments produced (P1 on the left and Mb top-down bottom)

No charge reduction/stripping



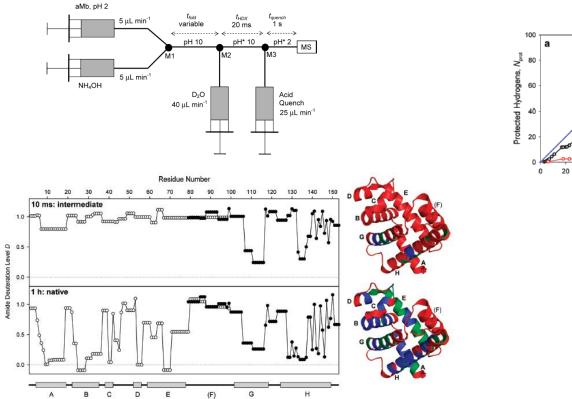


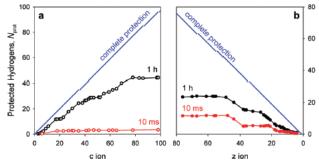
Top-Down HDX-MS



Number of reports still limited (C.H. Borchers, L. Konermann, I.A. Kaltashov,...).

Mostly on smaller, well behaving, well characterized proteins - e.g. Mb folding

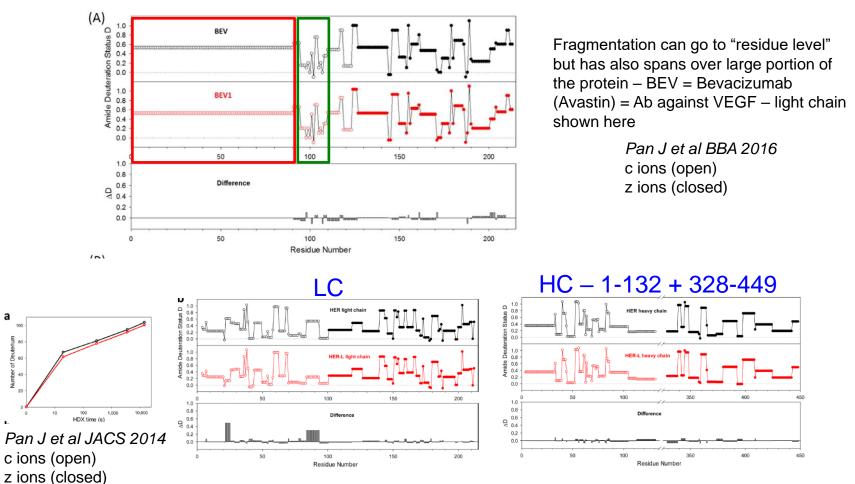




Pan J et al Anal Chem 2010 c ions (open) z ions (closed)

Top-Down HDX-MS

Ab studies also reported, however, Ab (150kDa) is not a big protein (HC 50kDa, LC 24kDa) + HC is "not interesting"





а

100

Number of Deuterium

Top-Down HDX-MS

Problem with ionization/fragmentation of some/larger proteins, not likely applicable (yet) to complexes.

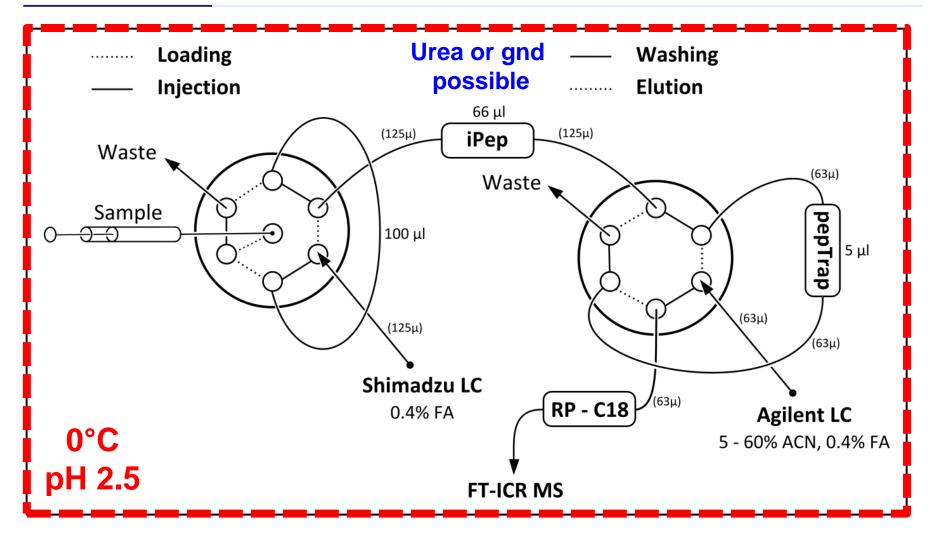
Poor fragmentation yield - coverage/resolution. Fragments mostly at the termini.

Motivation/advantages – sequence coverage sometimes not complete – already two complementary fragment ions do the job!

Digestion is often not easy, more handling=bigger D-loss (BE in bottom-up 15-35%, in topdown approx 2% using sub-zero LC – Pan J et al. JACS 2014)



H/D exchange workflow





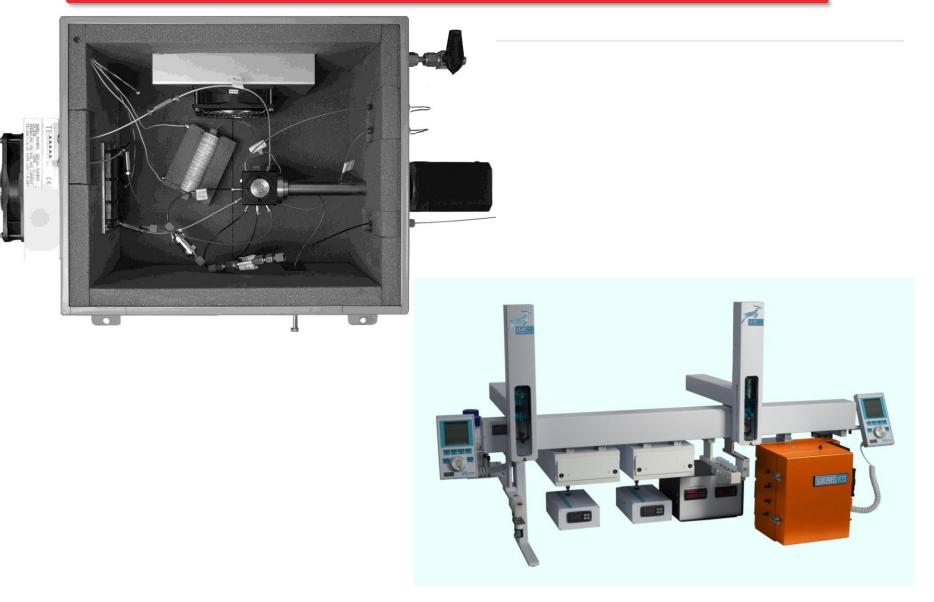
LAB SETUP



KEEP CALM AND PIPETTE ON



LAB SETUP





LAB SETUP





Focus on: Proteolysis

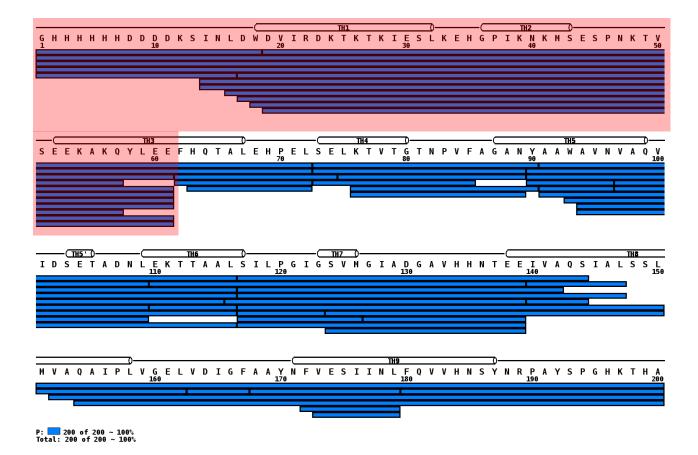
- defines spatial resolution of the method / full sequence coverage wanted
- Golden standard porcine pepsin A not all proteins can be efficiently by pepsin digested under HDX-MS conditions
- poor choice of commercial proteases suitable for HDX-MS type XIII – Aspergillopepsin, type XVIII – Rhizopuspepsin – both not very well defined crude extracts
- Immobilized protease high local protein-enzyme ratio, tunable via column size, flow, pressure, temperature



T-domain - proteolysis

Proteolysis

- defines spatial resolution of the method
- not all proteins can be efficiently by pepsin digested under HX-MS conditions

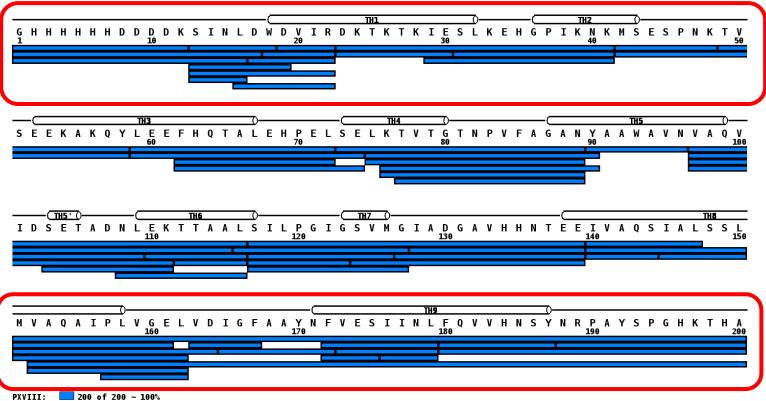




T-domain - proteolysis

Try different proteases (commercial extracts – protease type XIII, type XVIII)

Digestion by pepsin+rhizopuspepsin (type XVIII)

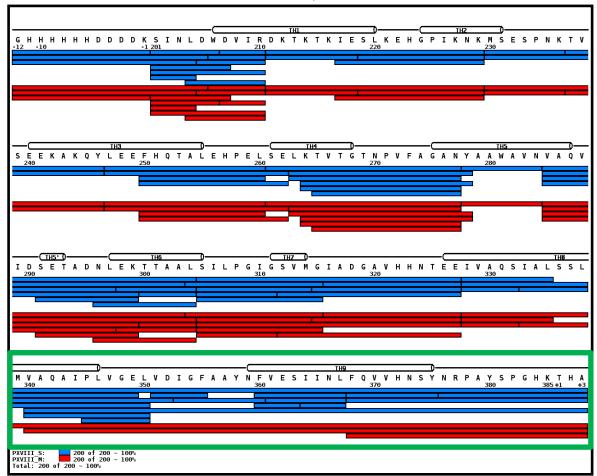


Total: 200 of 200 ~ 100%



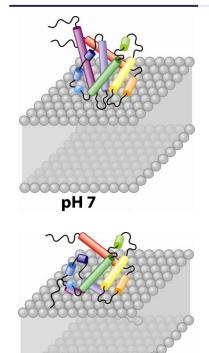
T-domain - mapping

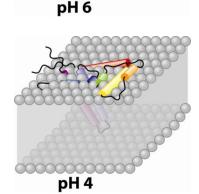
T-domain in solution , T-domain+LUVs





Summary for T-domain interacting with the membrane





pH 7 to pH 6 – still close to the native fold with just TH8 and TH9 inserting into the membrane (membrane bound state)

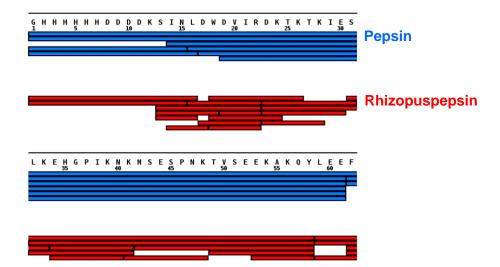
pH 6 to pH 4 – melting of the structure. TH8-9 deeply inserted, stabilized by TH5 and TH5'.

N-terminal part (mainly TH1 and TH3) stabilizes the membrane inserted state via ionic interactions.



Acidic proteases in HX – overview

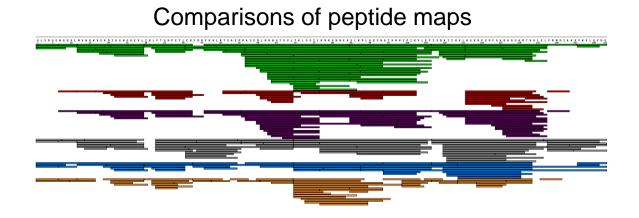
- 2017 Tsiatsiani et al An-PEP + Nirudodhi et al dual proteases (XIII+pepsin)
- 2016 Rey et al neprosin (not yet for H/D)
- 2015 Yang et al nepenthesin-2
- 2014 Kadek et al nepenthesin-1
- 2013 Ahn et al rice field eel
 2013 Rey et al nepenthes pitcher fluid
- 2010 Marcoux et al plasmepsin 2
- 2009 Rey et al recombinant rhizopuspepsin (protease type XVIII)
- 2007 Brier et al pepsins from Antarctic rock cod
- 2003 Cravello et al protease type XIII and XVIII Dykstra et al – endothiapepsin
- 2002 Wang et al pepsin immobilization
- 1993 Zhang&Smith MS + protein digestion (pepsin)



1979 - Rosa&Richards – protein digestion (pepsin, rhizopuspepsin)

Proteases – what is available (in our lab)

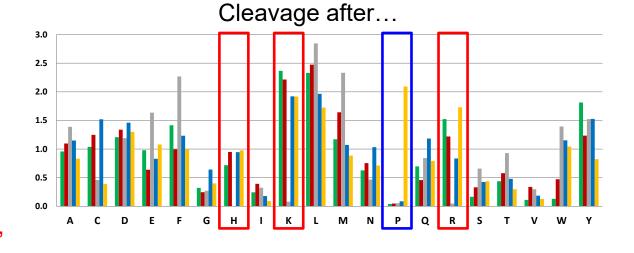
Nepenthesin-1 Nepenthesin-2 Rhizopuspepsin (XVIII) Aspergillopepsin (XIII) Pepsin AN-PEP Orzyasin Neprosin



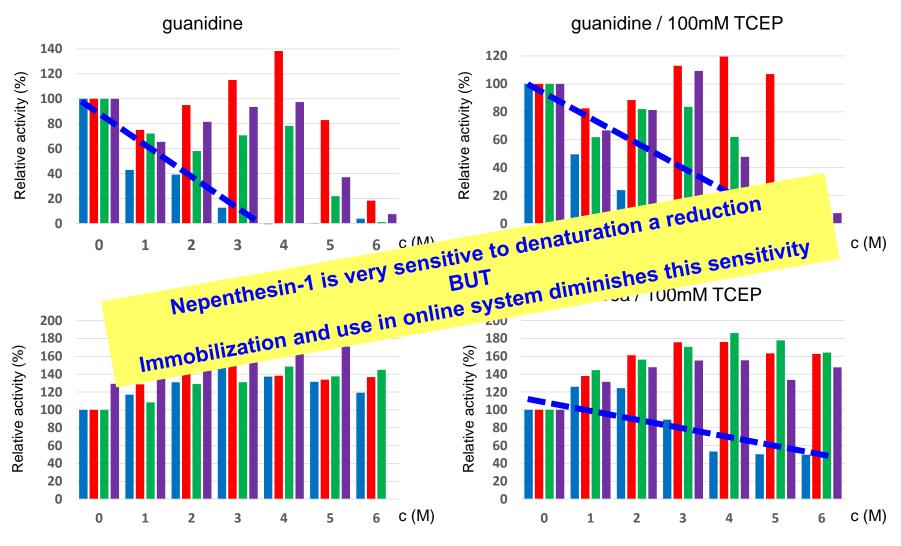
GOALS

Digest every protein

Care about average peptide length, redundancy, reproducibility, suitability for ETD,...



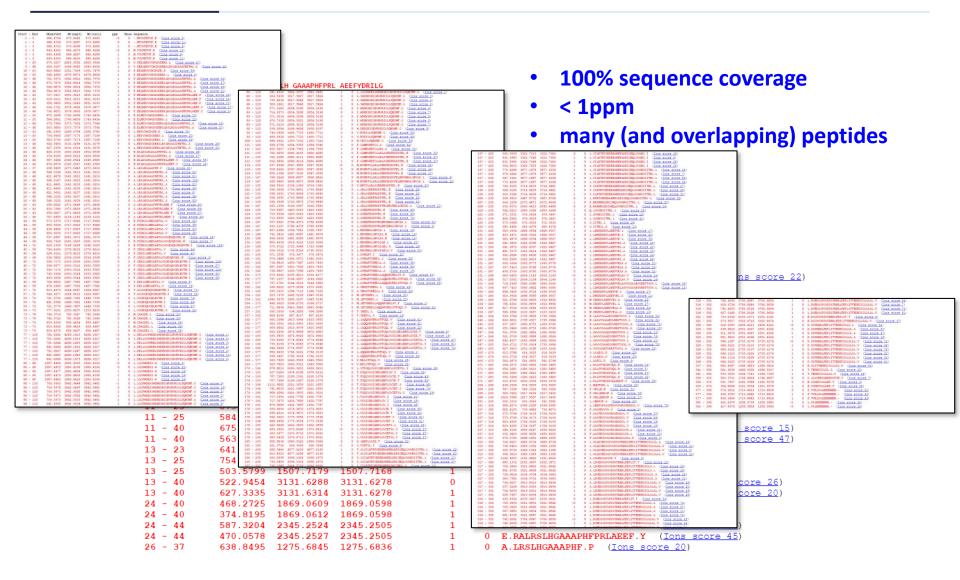
Proteases – effect of denaturing and reducing agents in solution



Nepenthesin-1, Pepsin, Nepenthesin-2, Rhizopuspepsin

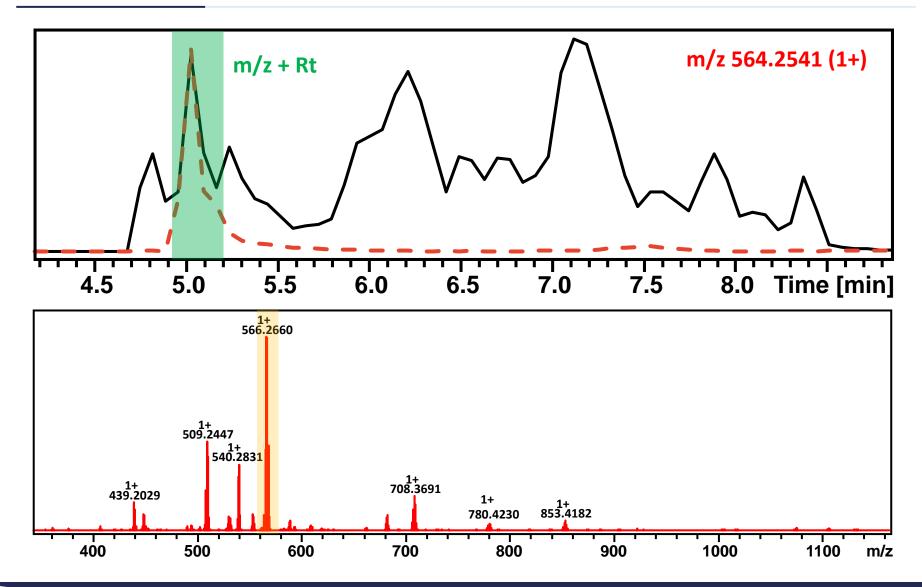


Focus on: Data interpretation





Focus on: Data interpretation – manual way





Focus on: Data interpretation – summary on the manual approach

Locating a peptide	30 s	
Plotting XIC	15 s	
Averaging a spectrum	10 s	
Export	20 s	
Opening file	15 s	(MagTran or mMass)
Deconvolution	10 s	
Writing down to xls	10 s	

2-10min per peptide from one state

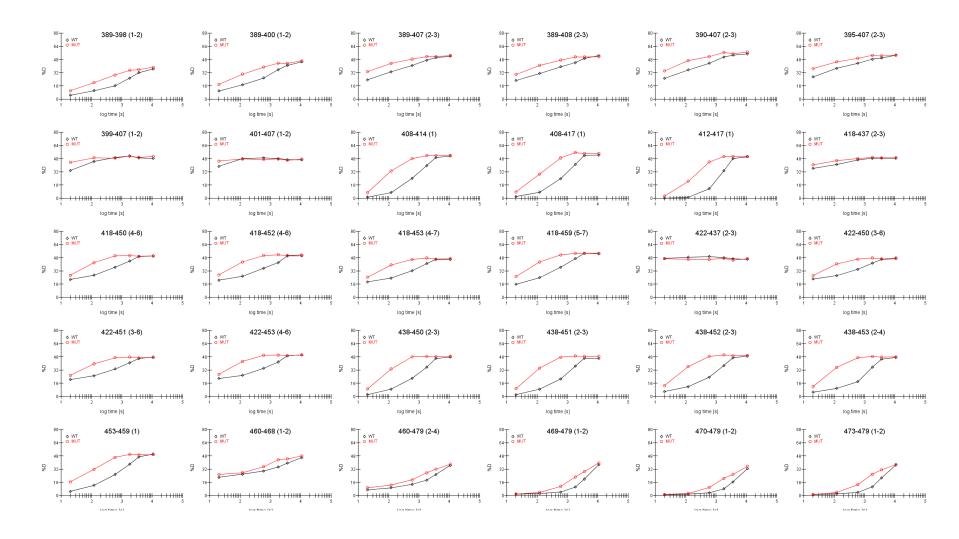
Software tools

Waters – DynamX LEAP / Sierra Analytics – HDEXaminer OmicsX – HDX Workbench

Mike Guttmann – HX Express (Excel base for Waters users, EX1/EX2 deconvolution)

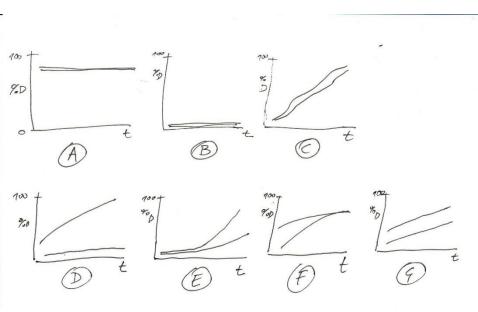


Output – uptake plots





Output – uptake plots



A: fast deuteration and remains constant - fast exchanging region, most likely no secondary structure

B: most probably deep in the structure - virtually no exchange. <u>Can be also fast exchanger</u>

C: structured region - deuterium uptake plots are somehow evolving

D-G: Various types of differences in deuteration

D: huge difference that appears already in the beginning and is getting even bigger during the exchange - clear difference is accessibility and H-bonding + signature of extremely stable binding of a ligand - the lower curve simply does not change while the upper one (for free protein) evolves

E: in contrast to D, this is poorly exchanging region with stable change upon ligand binding - it takes some time for the differences to appear but once they are there, the curves are separating

F: ligand binding occurs but is not so stable. so during the time of exchange we can observe dissociation events during which the protein has chance to gets deuterated.

G: stable change, similar to D



Back-exchange correction

Correction for back-exchange Mp-Mn Corr %D = _____*100 Mf-Mn

Corr No D = _____ *Ns Mf-Mn

Mp - MW partially deuterated Mn - MW nondeuterated Mf - MW fully deuterated (equilibrium) Ns - number of exchangeable sites (Σamide bonds-Pro) Important for comparison of different sequences, mutants

Not needed in typical comparison experiment

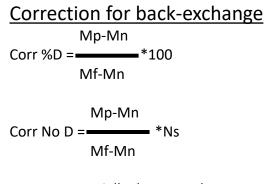
Needed when conclusions about protein folding or fold are made – helps to distinguish fast exchanging (in- as well as out- exchangers) and poor accessibility/strong hydrogen bodning

Protein incubated in D2O (under denaturing, heating,... conditions) and analyzed like the samples. Alternative – pre-digestion and deuteration of peptides!

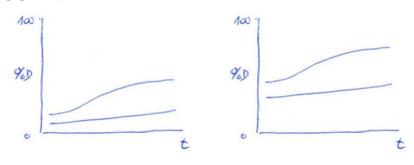
In general between 15-25%



Back-exchange correction

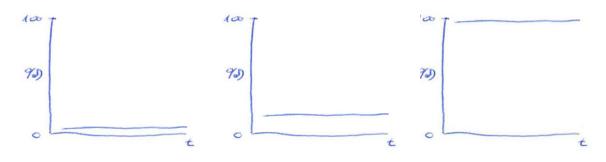


Mp - MW partially deuterated Mn - MW nondeuterated Mf - MW fully deuterated (equilibrium) Ns - number of exchangeable sites (Σamide bonds-Pro) Examples – before (left) and after (right) correction to BE – overall level raises + difference is slightly bigger (due to bigger y-scale)



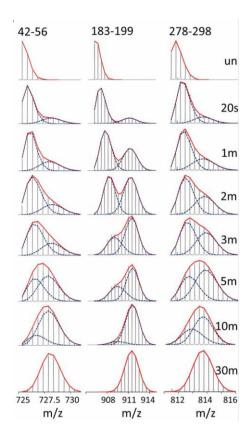
Examples – slow and fast exchanger

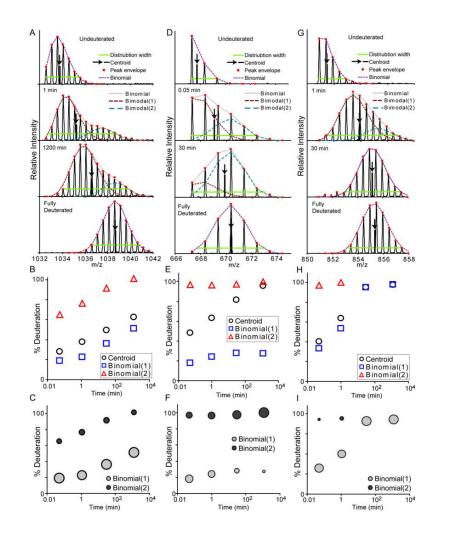
left – before BE correction hard to judge – middle – slow exchanger (deep in the structure) after correction; right – fast exchanger – e.g. HisTag - after correction





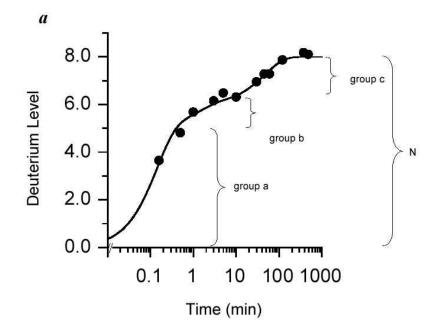
EX1/EX2 analysis







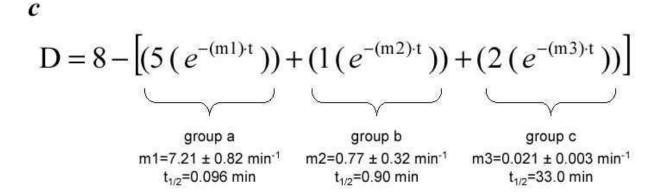
Datamining



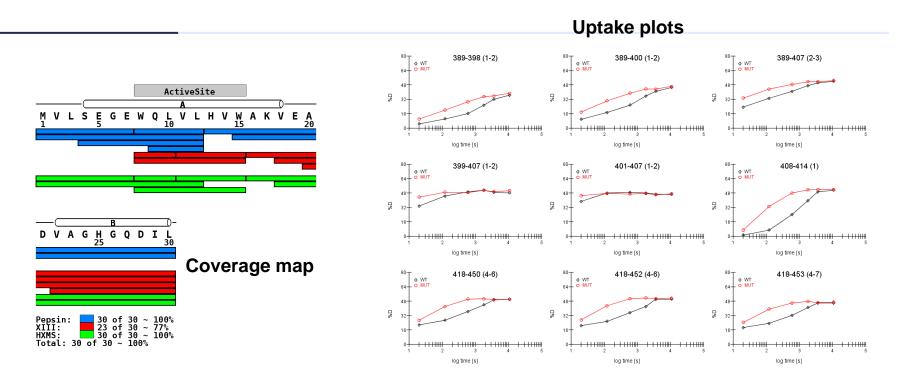
If fine sampling in HDX is used, the curve can be fitted and number of fast, slow and intermediate exchanging amides can be deduced

b

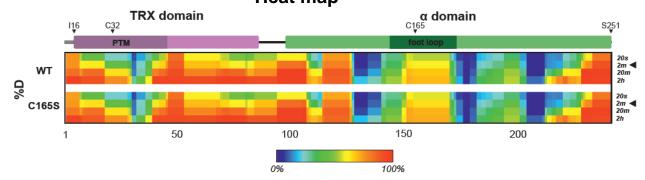
 $D = N - \sum_{i=1}^{N} \exp^{-k_i t}$



Data presentation

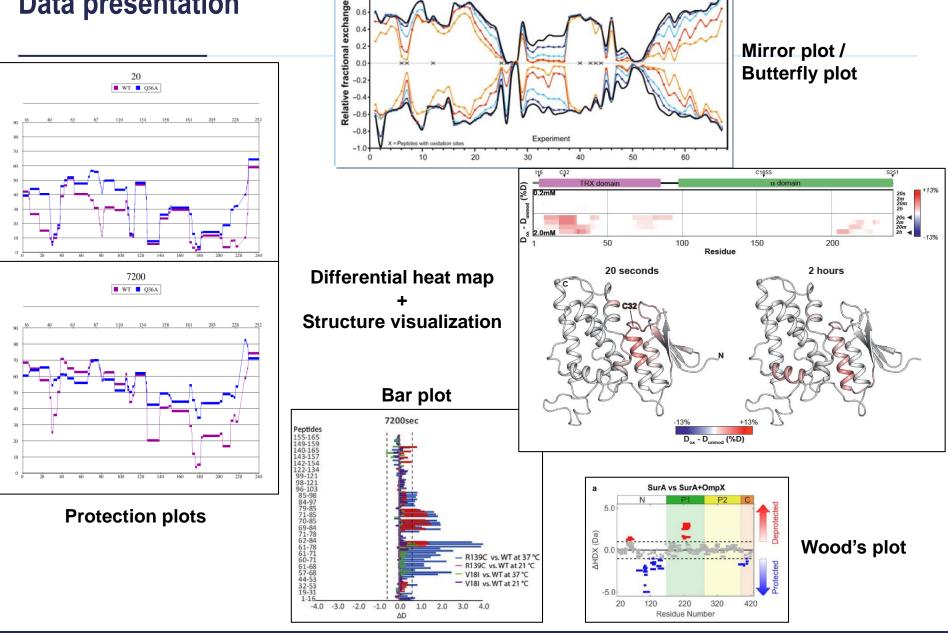


Heat map





Data presentation



Reference

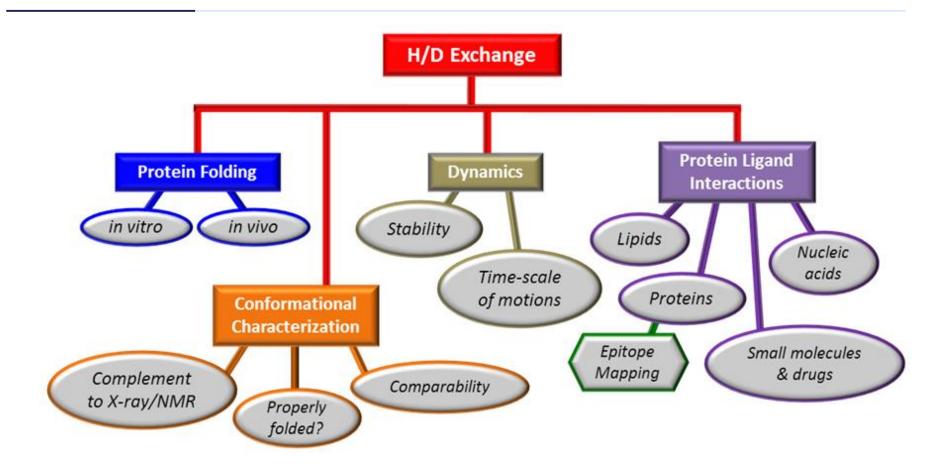
1.0

0.8

(a)



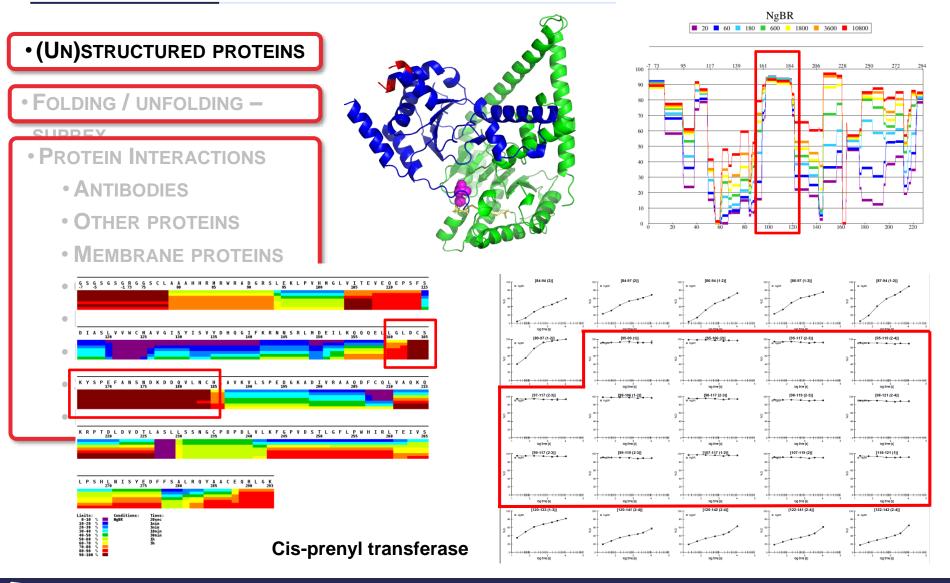
WHAT CAN BE DONE



Engen et al. (2011). Ency. Anal. Chem. ISBN 978-04-700-2731.

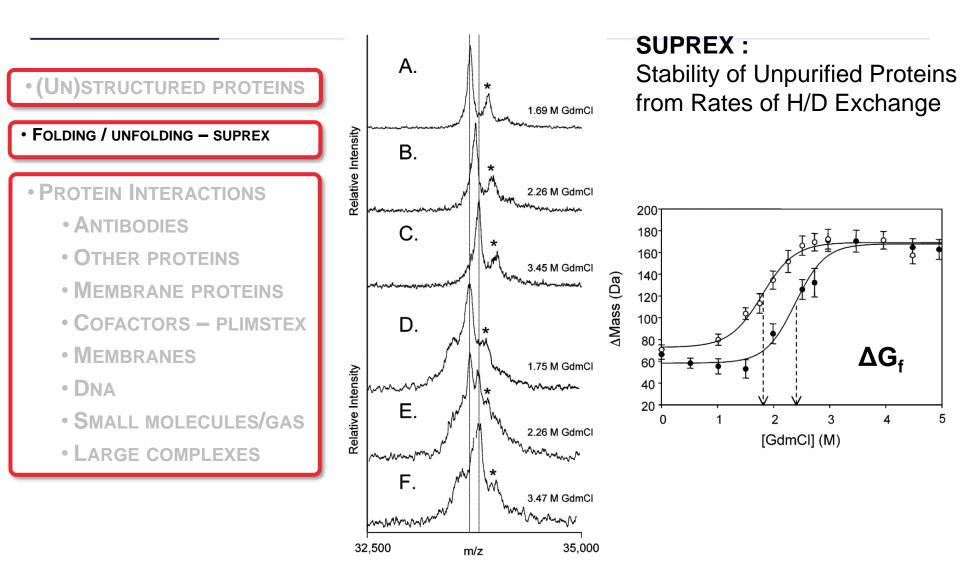


PROTEIN FOLD – STRUCTURED REGIONS





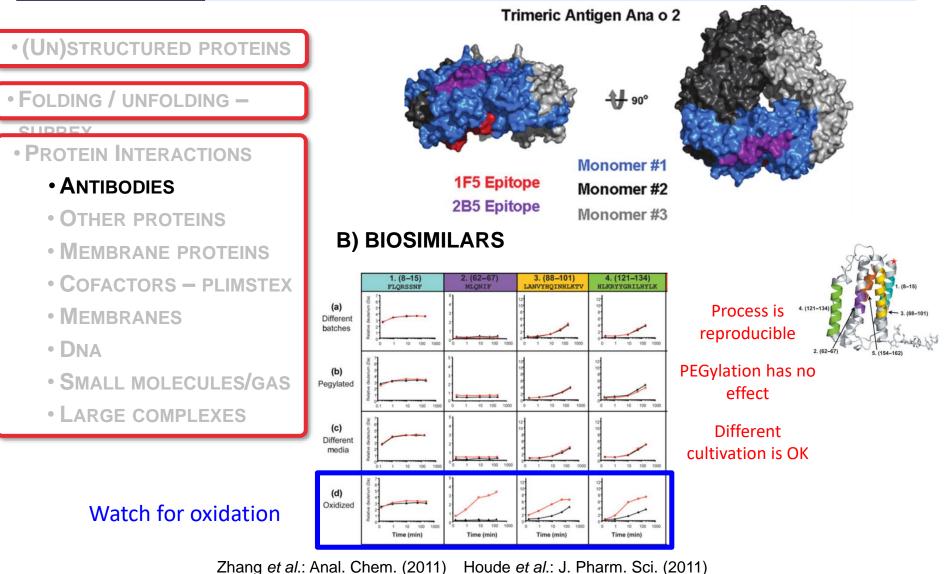
PROTEIN FOLDING





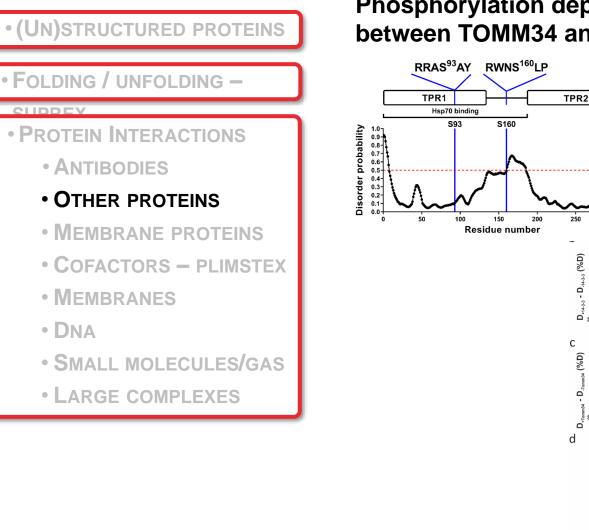
BIOLOGICALS

A) EPITOPE MAPPING

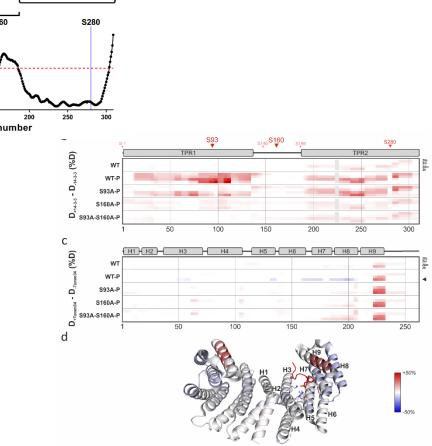


15

PROTEIN-PROTEIN INTERACTIONS



Phosphorylation dependent interaction between TOMM34 and 14-3-3





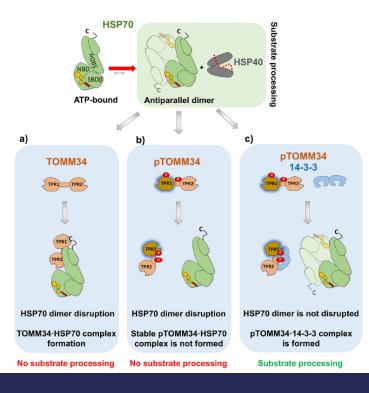
PROTEIN-PROTEIN INTERACTIONS

• (UN)STRUCTURED PROTEINS

• FOLDING / UNFOLDING – SUPREX

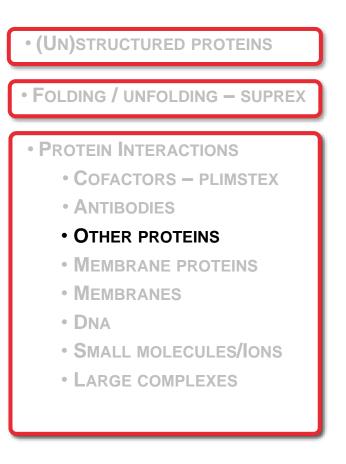
- PROTEIN INTERACTIONS
 - COFACTORS PLIMSTEX
 - ANTIBODIES
 - OTHER PROTEINS
 - MEMBRANE PROTEINS
 - MEMBRANES
 - DNA
 - SMALL MOLECULES/IONS
 - LARGE COMPLEXES

Phosphorylated TOMM34 interacts with 14-3-3 through its unstructured linker (no change in HDX detected) but it leads to overall structure "opening". Weak protection is observed on14-3-3.

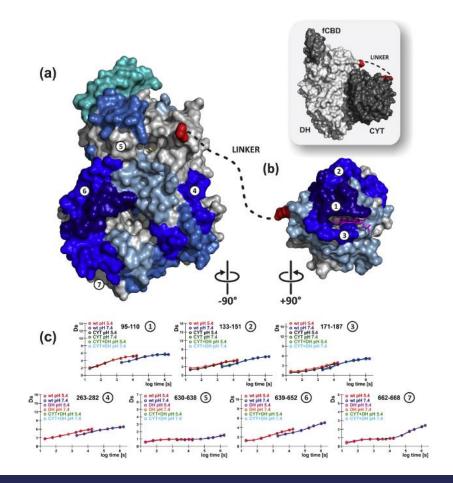




PROTEIN-PROTEIN INTERACTIONS



Cellobiose dehydrogenase – pH dependent intraprotein domain-domain interaction = protection. But it is all different!



Kadek A et al BBA 2017

M5

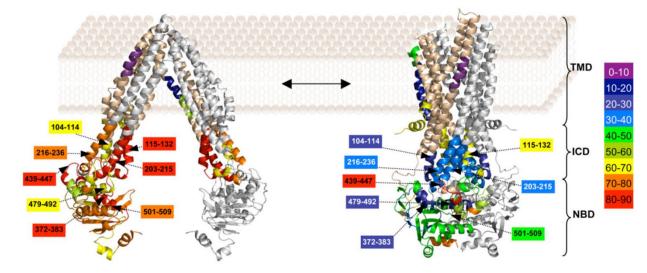
MEMBRANE PROTEINS

• (NE)STRUKTUROVANÉ

PROTEINV

• FOLDING / UNFOLDING -

- **SUPREX**
- PROTEIN INTERACTIONS
 - ANTIBODIES
 - OTHER PROTEINS
 - MEMBRANE PROTEINS
 - COFACTORS PLIMSTEX
 - MEMBRANES
 - DNA
 - SMALL MOLECULES/GAS
 - LARGE COMPLEXES



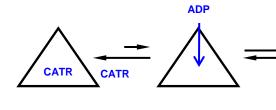


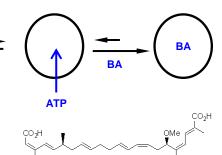


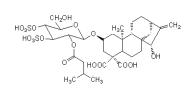
MEMBRANE PROTEINS IN NON-IONIC DETERGENTS

ANT, ADP/ATP TRANSPORTER









X-RAY STRUCTURE KNOWN FOR CATR

IMPOSSIBLE TO GET STRUCTURE FOR BA

REQUIRES PRESENCE OF DETERGENT (TRITON X-100)

C σ ٩ Ε Ð Ε

ATP out

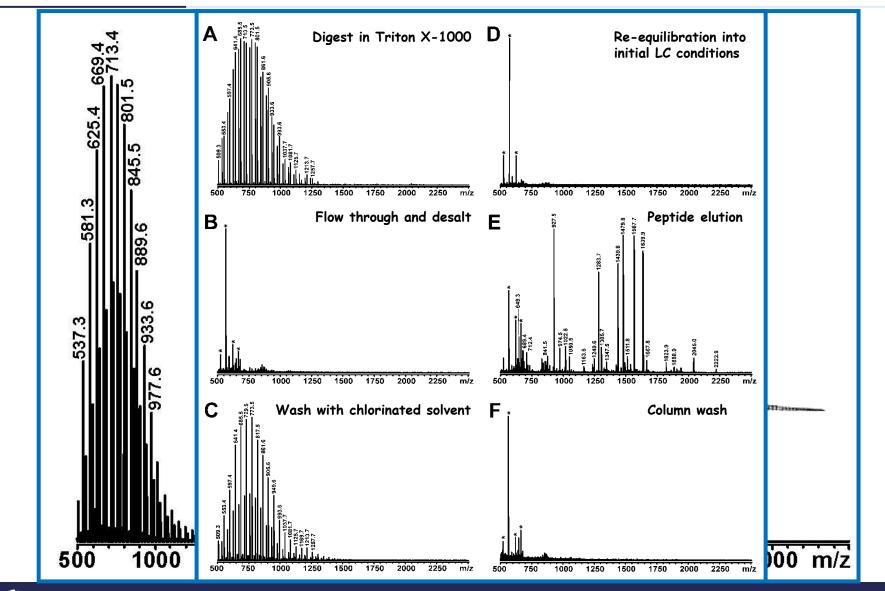
Ð

<u>ــ</u>

ADP in

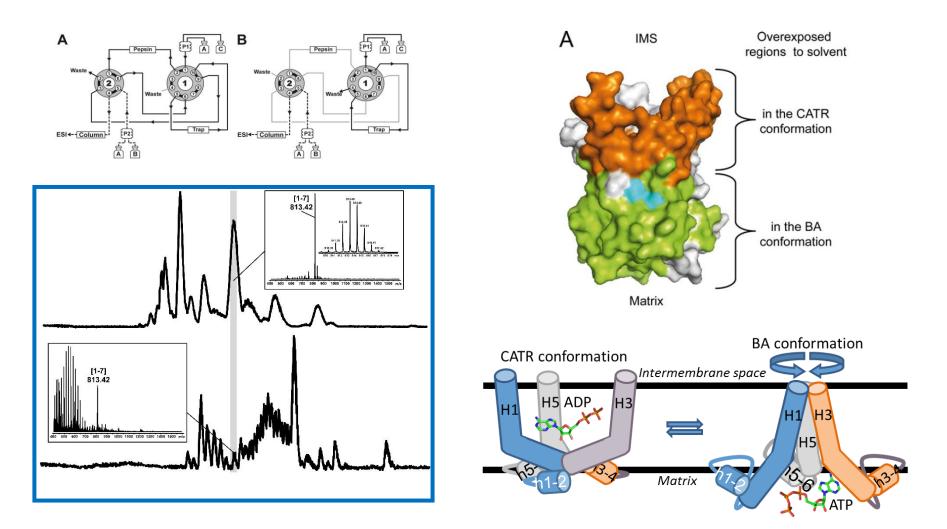


MEMBRANE PROTEINS IN NON-IONIC DETERGENTS



Rey M. et al. Anal. Chem. 2010

MEMBRANE PROTEINS IN NON-IONIC DETERGENTS

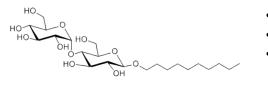




Rey, M. et al. J. Biol. Chem. 2010

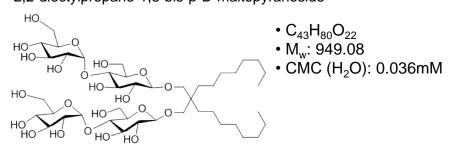
MEMBRANE PROTEINS IN ALKYLGLYSIDES

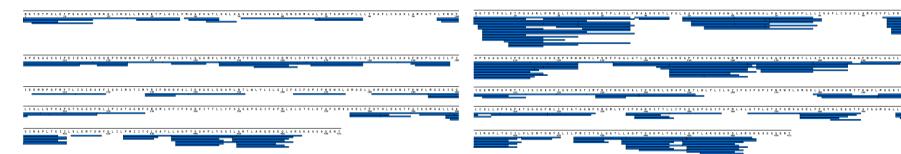
DM – Decyl Maltoside n-decyl-β-D-maltopyranoside

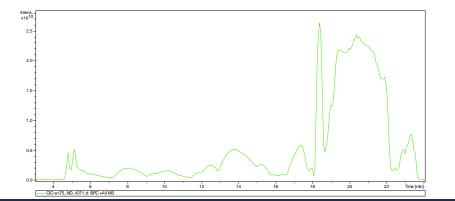


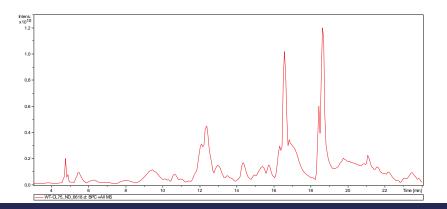
- C₂₂H₄₂O₁₁ • M_w: 482.6
- CMC (H₂O): 1.8mM

DMNG – Decyl Maltose Neopentyl Glycol 2,2-dioctylpropane-1,3-bis-β-D-maltopyranoside





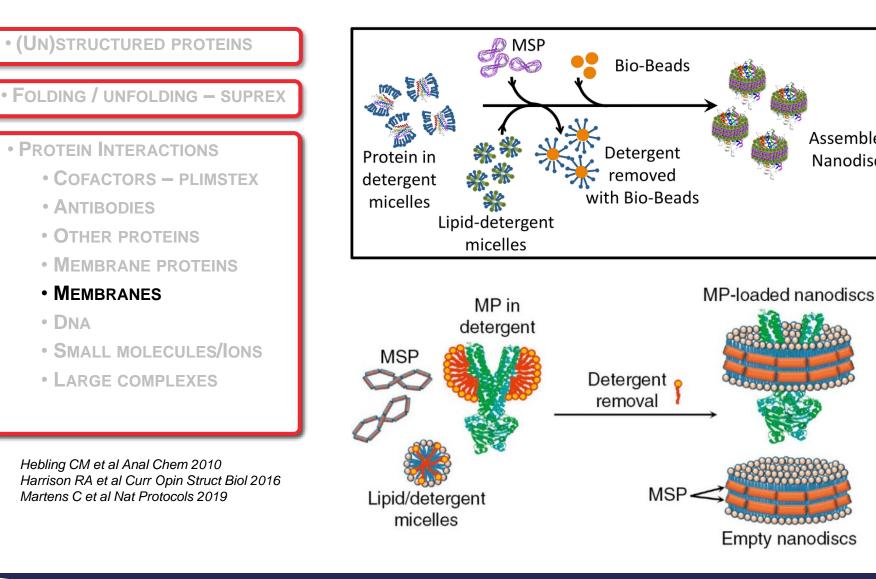




MEMBRANE PROTEINS IN NANODISCS

Assembled

Nanodiscs





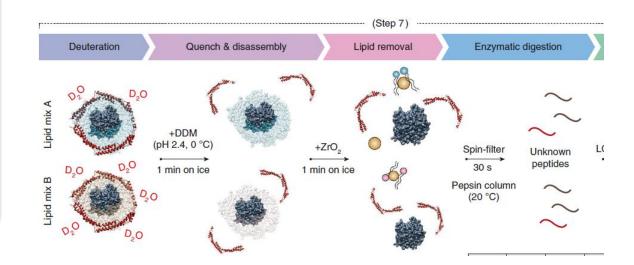
MEMBRANE PROTEINS IN NANODISCS

• (UN)STRUCTURED PROTEINS

- FOLDING / UNFOLDING SUPREX
- PROTEIN INTERACTIONS
 - COFACTORS PLIMSTEX
 - ANTIBODIES
 - OTHER PROTEINS
 - MEMBRANE PROTEINS
 - MEMBRANES
 - DNA
 - SMALL MOLECULES/IONS
 - LARGE COMPLEXES

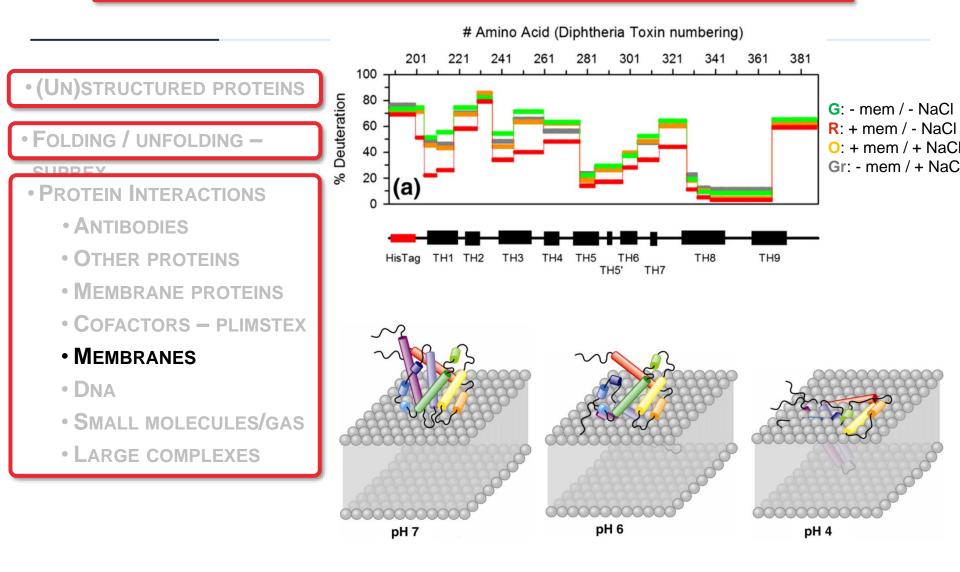
Nanodiscs – protocol (Martens C et al Nat Protocols 2019)

- Quench + disassembly (detergent)
- Lipid removal by ZrO2 coated resin
- ZrO2 particles filtering
- Digestions (eventually done during lipid removal)



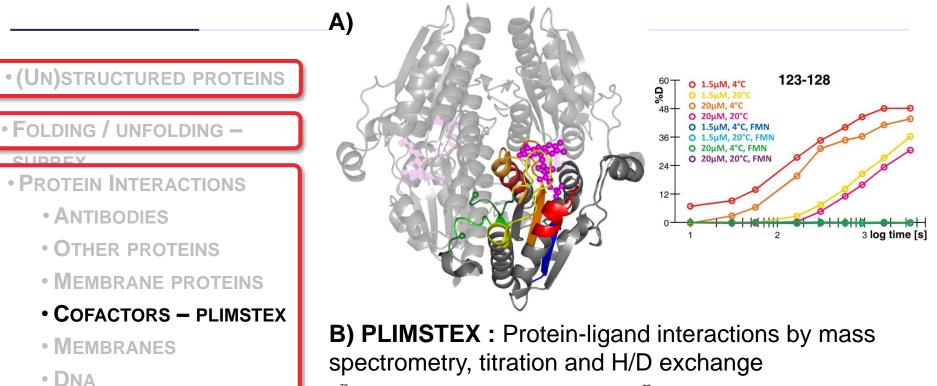


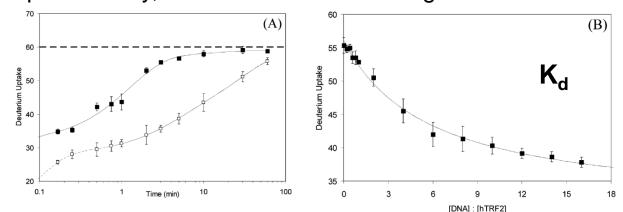
MEMBRANE INTERACTION





COFACTOR BINDING





Zhu et al.: JACS (2003) Sperry et al.: Biochemistry (2008)



• SMALL MOLECULES/GAS

LARGE COMPLEXES

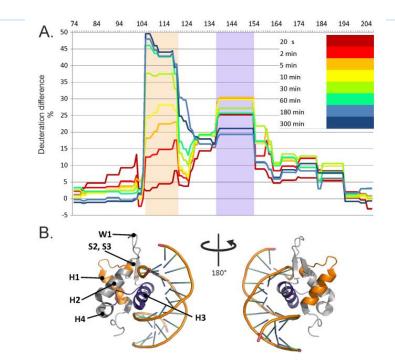
DNA BINDING



- PROTEIN INTERACTIONS
 - COFACTORS PLIMSTEX
 - ANTIBODIES
 - OTHER PROTEINS
 - MEMBRANE PROTEINS
 - MEMBRANES
 - DNA
 - SMALL MOLECULES/IONS
 - LARGE COMPLEXES

Slavata et al.: Biomolecules 2019

Sperry J.B. et al J Am Soc Mass Spectrom 2008 Poliakov A. et al Rapid Commun Mass Spectrom 2008 Roberts V.A. et al Nucleic Acid Res 2012 Graham B.W. et al J Biol Chem 2016



DNA ISSUES!!!

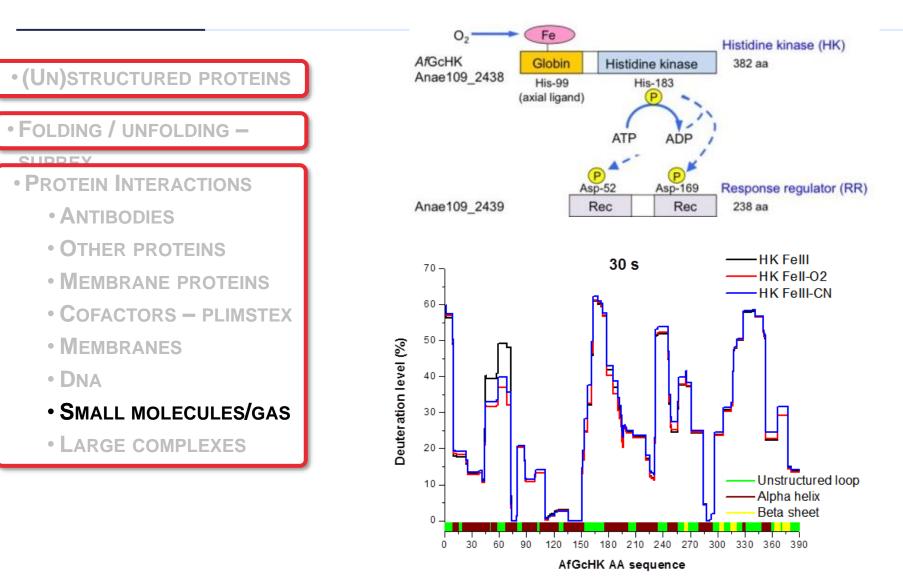
DNA gets protonated upon quenching/acidification and may precipitate. Protein co-precipitate. DNA affects LC separation on RP columns

Tricks available (published)

- Strong anion exchange (DNA removal)
- Charged compounds added (protamine)
- Denaturation
- LC system washing (neutral pH+organics)



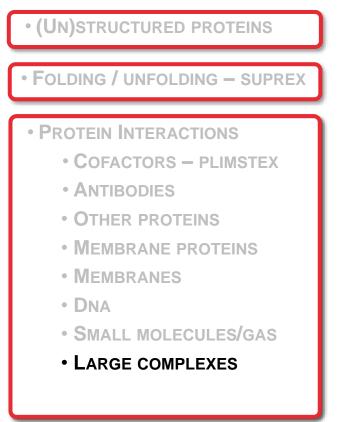
SMALL MOLECULES / GAS SENSING

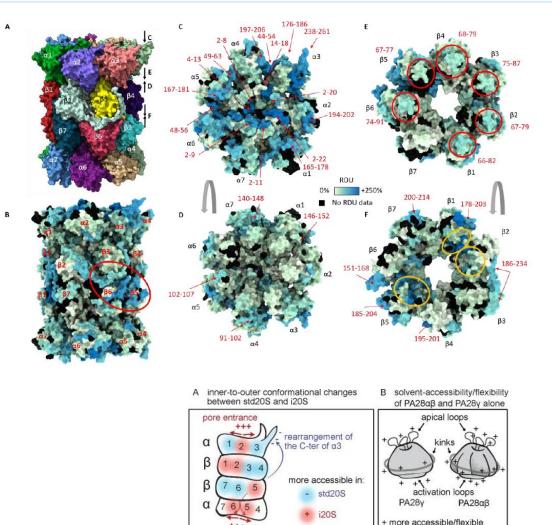


Martinkova, Kitanishi, Shimizu: J. Biol. Chem. (2013)



SUPRAMOLECULAR COMPLEXES

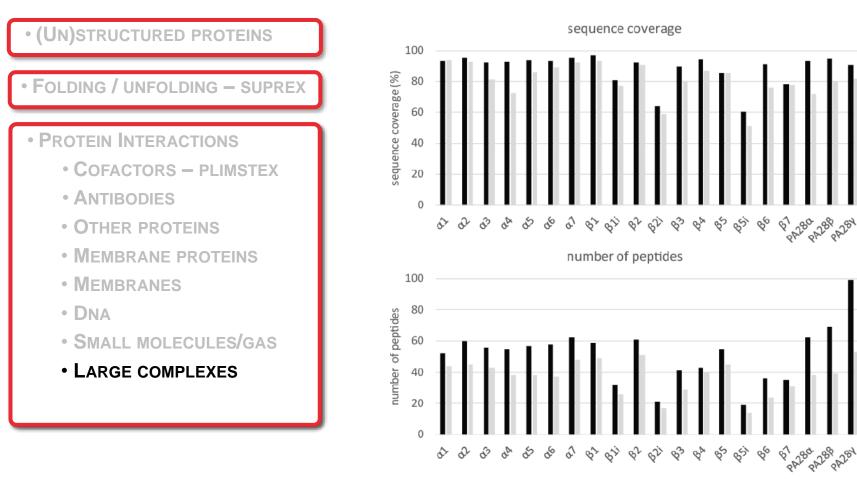




Lesne J et al bioRXiv 2020



SUPRAMOLECULAR COMPLEXES

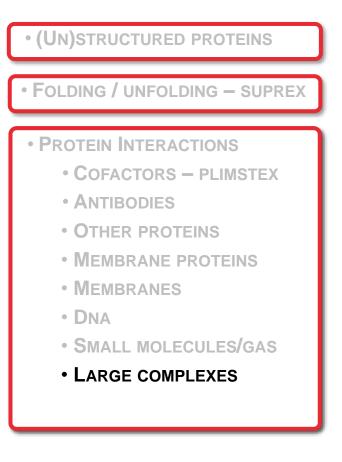


Lesne J et al bioRXiv 2020

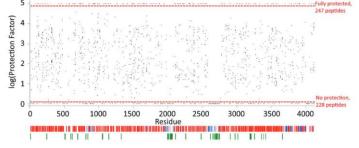




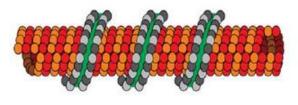
HUGE PROTEINS



Sheff JG et al Analyst 2017 – DNA-PKc (469kDa)



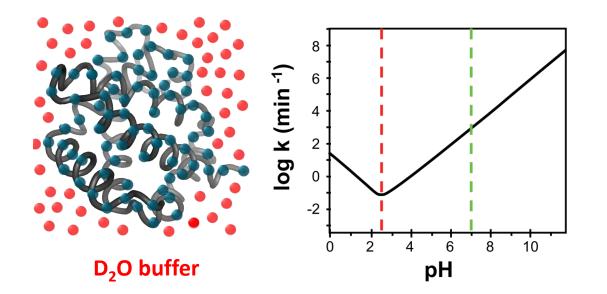
Burns KM et al Prot Sci 2015 – alpha/beta tubulin + dimeric kinesin 175kDa total sequence assembled >1GDa







Pros and cons of HDX-MS



$$k(T_2) = k(T_1)exp^{-\frac{E_a}{R}(\frac{1}{T_2} - \frac{1}{T_1})}$$



Buffer – can be whatever
Protein size – does not matter
Protein concentration – can be very low
Mixture compatible – many proteins
Temperature or pH – can be whatever
(correction must be done - H/D depends on pH and T)
Commercial solutions available

Dilutions – into D_2O and to lower the pH **Protein concentration** – not all complexes are formed at given concentration (K_D) **Comparison** – at least two states required

